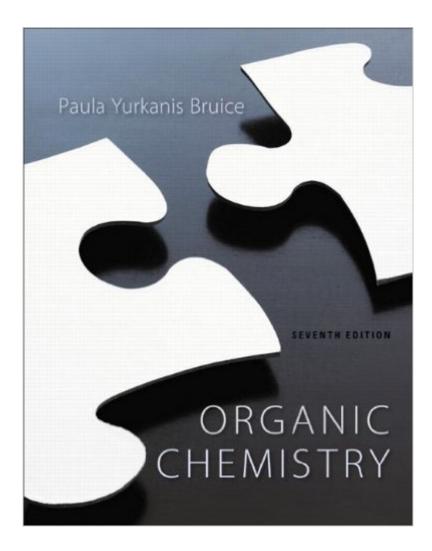
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Organic Chemistry (7th Edition)





Synopsis

All of Paula Bruiceâ TMs extensive revisions to the Seventh Edition of Organic Chemistry follow a central guiding principle: support what modern students need in order to understand and retain what they learn in organic chemistry for successful futures in industry, research, and medicine. In consideration of todayâ TMs classroom dynamics and the changes coming to the 2015 MCAT, this revision offers a completely new design with enhanced art throughout, reorganization of materials to reinforce fundamental skills and facilitate more efficient studying. Â

Book Information

Hardcover: 1392 pages Publisher: Pearson; 7 edition (January 8, 2013) Language: English ISBN-10: 0321803221 ISBN-13: 978-0321803221 Product Dimensions: 8.7 x 1.9 x 11 inches Shipping Weight: 6.4 pounds (View shipping rates and policies) Average Customer Review: 3.3 out of 5 stars Â See all reviews (43 customer reviews) Best Sellers Rank: #6,632 in Books (See Top 100 in Books) #6 in Books > Science & Math > Chemistry > Organic #20 in Books > Textbooks > Science & Mathematics > Chemistry #1982 in Books > Reference

Customer Reviews

This is by far the worst organic chemistry textbook I have used, and is probably one of the worst textbooks I have ever come across. Aside from those who thought that placing their reviews under the product page was the same thing as rating the seller, I have no idea why there are so many positive reviews for it. It is excessively verbose, often confusingly worded, and generally ordered in a very discontinuous, confusing way. Really the only good thing about the ordering is that Bruice discusses stereochemistry near the beginning of the book (chapter 4), prior to even discussing any of the reactions, while other books tend to address it much later. By tackling it it early, students are much better able to understand how the molecules take shape after the reactions occur. Unfortunately, many other important topics are oddly disjointed and poorly presented. For example, she discusses the criteria for aromaticity and Huckel's rule in chapter 8, but then doesn't actually go into the reactions, like the Grignard reagents, but Bruice does not explain their

extremely important interactions with carbonyl compounds until chapters 16 and 17, once she finally gets around to the carbonyls. And the chapters on carbonyls, 16, 17, and 18, are probably the worst in the book, in my opinion. Rather than starting with a basic overview of the 4 or 5 general reaction types, she starts straight with nucleophylic acyl substitution and carbonyl derivatives, such as acyl chlorides, esters, etc. She only gets to the aldehydes and ketones in the next chapter. She does loosely group the chapters by reaction type but somehow the overall effect is a messy jumble of multiple functional groups.

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